

Applicability of UV/H₂O₂-cum-VacuumUV/H₂O advanced oxidation process for degradation of micropollutants in different water matrices

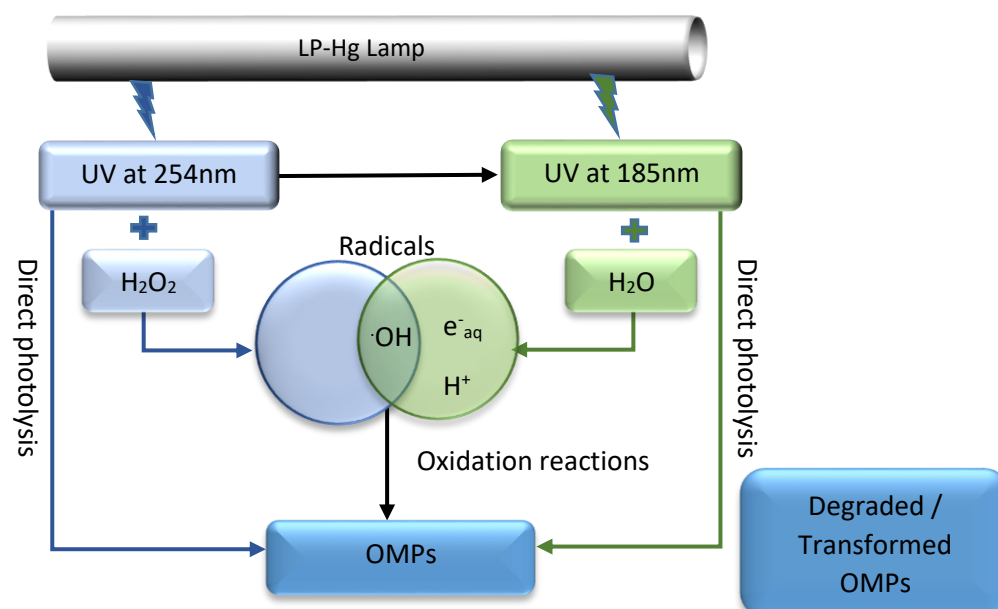
Areas of research: Advanced oxidation process; Micropollutants; Transformation by-products; (Photo)chemistry
 Duration; start date; allowance: min.3 months; November 2019; 350€/month

Background

Did you know that sometimes up to 90% of the drug that you swallow is excreted? What happens to the shampoo you just lathered your hair with or the remains of the detergent that washes your clothes fresh? Once, they do their job they end up in our waters and are called Organic micro-pollutants (OMP's). Since pharmaceuticals are toxic, persistent and bio accumulative and are designed to act at lower concentrations, it is not surprising that even low concentrations reaching the environment can be detrimental. For example, on exposure of contraceptive pills, male fish may feminize thus affecting their population. OMPs are now increasingly found in the water sources and we are committed to reduce any health impact it may cause. There are ten thousands of such OMPs released to the environment. One of the most popular approach to tackle the issue is to have an Advanced Oxidation Processes (AOP). AOPs are water treatment processes that produces and further uses the hydroxyl radicals ($\cdot\text{OH}$), one of the strongest, highly reactive and (mostly) non-selective oxidant, for transformation or degradation of OMPs. One of the most widely used AOP is the UV_{254nm}/H₂O₂ where the UV radiation at 254nm (UV_{254nm}) photo-chemically splits the hydrogen peroxide (H₂O₂) molecule to produce $\cdot\text{OH}$.

Project description

The UV_{254nm}/H₂O₂ process requires H₂O₂ to be added as an external oxidant, for the production of $\cdot\text{OH}$, complicating the logistics of the entire process and increasing the cost of the process since any unused H₂O₂ has to be quenched. In this project, we combine this process with Vacuum UV radiation at 185nm (VUV_{185nm}) that can directly split the H₂O molecule. Added advantages are that: the VUV_{185nm}/H₂O produces more like the solvated electrons (e^-_{aq}) and the protons (H⁺) helping to degrade a wider range of OMPs alongside the fact that the lamp that produces both UV_{254nm} and VUV_{185nm}, without any no extra energy input.



Objective of the MSc. Thesis/Internship

Your part in this project would be to conduct photochemical degradation experiments under the irradiation of dual wavelength using a collimated beam setup (that houses the lamp producing the radiation) on selected OMPs and study the effect of the following parameters on the degradation kinetics:

1. pH
2. Temperature
3. Scavenger (organic matrix)
4. Inorganic matrix

As we are also working towards developing a pilot-scale, you could potentially be part of this designing realized using the commercial computational fluid dynamics software, Ansys Fluent. The designing requires combined knowledge of chemical reactor design and photochemistry.

Requirements

1. Motivated, enthusiastic, team-spirited and independent
2. Experience in laboratory
3. Proficiency in English
4. EU/Non-EU student enrolled in any university within The Netherlands or EU student enrolled in universities elsewhere in the EU. Unfortunately, non-EU students enrolled outside the Netherlands cannot apply.

Application

Did this project instil interest in you? Do you want to know more to make a decision? Or do you want to apply right away? Contact me at nimmy.kovoor@wetsus.nl (to apply, please attach a motivation letter of 1 page and a CV of not more than 2 pages). To know more about the topic, please go through the reference papers.

References

1. Gonzalez, M., Oliveros, E., Worner, M. and Braun, A. (2004). Vacuum-ultraviolet photolysis of aqueous reaction systems. *Journal of Photochemistry and Photobiology C: Photochemistry Reviews*, 5(3), pp.225-246.
2. Zoschke, K., Börnick, H. and Worch, E. (2014). Vacuum-UV radiation at 185 nm in water treatment – A review. *Water Research*, 52, pp.131-145.